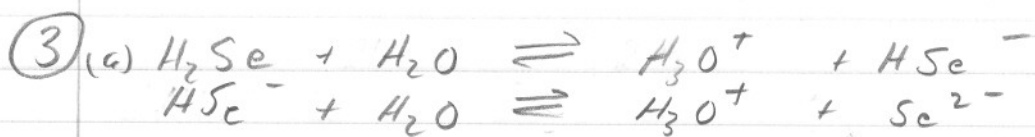
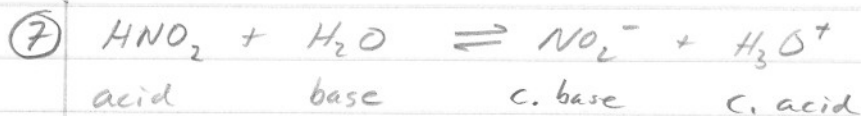


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⑥ CH_4 and H_2S
- there is the possibility of the compound losing a H^+ ion.



⑧ - it accepts protons (Brønsted-Lowry base)
- but it does not have an OH group (Arrhenius base)