

Table of Polyatomic Ions

Name	Formula
acetate	$\text{C}_2\text{H}_3\text{O}_2^-$ OR CH_3COO^-
ammonium	NH_4^+
bicarbonate (hydrogen carbonate)	HCO_3^-
bisulfate (hydrogen sulfate)	HSO_4^-
bisulfide (hydrogen sulfide)	HS^-
bisulfite (hydrogen sulfite)	HSO_3^-
bromate	BrO_3^-
carbonate	CO_3^{2-}
chlorate	ClO_3^-
chlorite	ClO_2^-
chromate	CrO_4^{2-}
cyanide	CN^-
dichromate	$(\text{CrO}_4)_2^{2-}$
dihydrogen phosphate	H_2PO_4^-
hydrogen oxalate	HC_2O_4^-
hydrogen phosphate	HPO_4^{2-}
hydroxide	OH^-
hypochlorite	ClO^-
hyponitrite	NO^-
hypophosphite	PO_2^{3-}
iodate	IO_3^-
nitrate	NO_3^-
nitrite	NO_2^-
oxalate	$\text{C}_2\text{O}_4^{2-}$
perchlorate	ClO_4^-
permanganate	MnO_4^-
pernitrate	NO_4^-
peroxide	O_2^{2-}
phosphate	PO_4^{3-}
phosphite	PO_3^{3-}
silicate	SiO_3^{2-}
sulfate (sulphate)	SO_4^{2-}
sulfite (sulphite)	SO_3^{2-}
thiocyanate	SCN^-
thiosulfate	$\text{S}_2\text{O}_3^{2-}$